1353

## Synthesis of the S-Enantiomer of Paniculidine A: Absolute R-Configuration of the Natural Paniculidines A and B

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The absolute *R*-configuration of both the title monochiral indole alkaloids has been established.

Monochiral prenylindoles paniculidine A (1) and paniculidine B (2) were recently isolated from the root bark of *Murraya* paniculata (Linn.) Jack<sup>1</sup> and were synthesized in racemic form.<sup>1,2</sup> Both alkaloids were shown to possess the same<sup>1</sup> but yet unknown absolute configuration. Here we report the preparation of chiral paniculidine A starting from the readily available (4*R*)-5-acetoxy-4-methylpentanoic acid (3).<sup>3</sup> The synthetic alkaloid was found to be the S-enantiomer of the natural compound. Hence, the natural paniculidines (1) and (2) belong to the *R*-series.

Coupling of the appropriate chiral C<sub>4</sub>-building block with a  $\alpha$ -formylindole fragment was chosen for the construction of the target molecule (see Scheme). For this purpose the acetoxy acid (3) was transformed into the homallylic acetate *R*-(4) using a procedure previously developed by us.<sup>4</sup> Its oxidative degradation using KMnO<sub>4</sub> on silica gel<sup>5</sup> gave the respective acetoxy



acid which was further converted, without any additional purification, into the known acetoxy ester (5).<sup>6</sup> Selective saponification of the latter furnished the known alcohol (6)<sup>7</sup> which, in turn, was smoothly transformed into the bromide (7): b.p. 56 °C/2 mmHg;  $n_D^{20}$  1.4550;  $[\alpha]_D^{23}$  + 16.7° (c 1.78, CHCl<sub>3</sub>). The chiral phosphonium bromide (8) was obtained in virtually



Scheme. Reagents and conditions: i, (a)  $KMnO_4/SiO_2$ ,  $C_6H_6$ , 25 °C, (b)  $CH_2N_2$ ,  $Et_2O$ , 25 °C (87% overall); ii,  $K_2CO_3MeOH$ , 0—5 °C (75%); iii,  $CBr_4$ , Py, Ph<sub>3</sub>P, THF, 25 °C (77%); iv, Ph<sub>3</sub>P, THF, 80 °C, 14 kBar (100%); v, LDA, N-tosyl-3-formylindole, THF–HMPA, -78 °C (60%); vi,  $H_2$ , 5% Pd–C, EtOH, 25 °C (100%); vii,  $C_{10}H_8/Na$ , DME, -60 °C (100%).

quantitative yield from (7) with the use of a high-pressure technique (cf. ref. 8); (8), m.p. 70–72 °C;  $[\alpha]_D^{24} + 27.3^{\circ}$  (c 1.78, CHCl<sub>3</sub>). The Wittig reaction of N-tosyl-3-formylindole<sup>9</sup> with the phosphorane generated from (8) gave smoothly the disubstituted *cis*-olefin (9) (>98% Z, <sup>1</sup>H n.m.r. data) as a colourless oil:  $[\alpha]_D^{25} + 82.7^{\circ}$  (c 1.64, CHCl<sub>3</sub>);  $\delta$ (CDCl<sub>3</sub>) 1.33 (d, J7 Hz, 3 H, CH<sub>3</sub>), 2.37 (s, 3 H, CH<sub>3</sub>), 3.64 (m, 1 H, CH), 3.78 (s, 3 H, CH<sub>3</sub>O), 5.82 (dd, J9 and 10 Hz, 1 H, HCCMe), 6.52 (d, J 10

Hz, HC=CCMe), 7.75 (s, 1 H, HCN), aromatic H: 7.2–7.4 (m, 4 H), indole H: 7.5–8.0 (m, 4 H). Catalytic hydrogenation of (9) yielded quantitatively the amido ester (10) as a colourless oil:  $[\alpha]_{D}^{27} + 8.7^{\circ}$  (c 1.64, CHCl<sub>3</sub>).

Finally, reductive cleavage of the N-S bond in (10) using sodium naphthalenide gave quantitatively (*ca.* 30% isolated yield for the above seven stages) the target compound as a colourless oil:  $[\alpha]_D^{25} + 31.3^\circ$  (*c* 0.45, CHCl<sub>3</sub>). For comparison,<sup>1</sup> the specific optical rotation of paniculidine A is  $[\alpha]_D^{24} - 31.9^\circ$  (*c* 0.1, CHCl<sub>3</sub>). Spectral properties (i.r., <sup>1</sup>H n.m.r., u.v.) of the compound prepared were found to be very close to those reported for the natural product.<sup>1</sup>

Hence, the compound thus synthesized from R-(4), without affecting its chiral centre, is represented by formula S-(1). Accordingly, the absolute R-configuration of the natural paniculidine A and paniculidine B is represented by formula R-(1) and R-(2), respectively.

## References

- 1 T. Kinoshita, S. Tatara, and U. Sankawa, *Chem. Pharm. Bull.*, 1985, **33**, 1770.
- 2 M. Somei and H. Ohnishi, Chem. Pharm. Bull., 1985, 33, 5147.
- 3 B. A. Czeskis and A. M. Moiseenkov, Khim.-Pharm. Zh., 1988, 597.
- 4 A. M. Moiseenkov and B. A. Czeskis, *Dokl. Akad. Nauk SSSR*, 1986, **290**, 1379.
- 5 J. T. B. Ferreira, W. O. Cruz, P. C. Vieira, and M. Yonashiro, J. Org. Chem., 1987, **52**, 3698.
- 6 T. Matsumoto, M. Takahashi, and Y. Kashihara, *Bull. Chem. Soc. Jpn.*, 1979, **52**, 3329; T. Matsumoto, S. Imai, S. Miuchi, and H. Sugibayashi, *ibid.*, 1985, **58**, 340.
- 7 Beilsteins 'Handbuch der Organischen Chemie,' Bd 3 (E IV), Springer-Verlag, Berlin, Heidelberg, New York, 1977, S.800; K. Mori and S. Senda, *Tetrahedron*, 1985, **41**, 541.
- 8 A. M. Moiseenkov, I. M. Zaks, and B. S. Elyanov, Zh. Obshch. Khim., 1983, 53, 1260; I. M. Zaks, B. S. Elyanov, V. M. Zhulin, and A. M. Moiseenkov, *Izv. Akad. Nauk SSSR*, Ser. Khim., 1984, 1094; W. G. Dauben, J. M. Gerdes, and R. A. Bunce, J. Org. Chem., 1984, 49, 4293.
- 9 D. D. Evans, Aust. J. Chem., 1973, 26, 2555; K.-F. Cheng, Y.-C. Kong, and T.-Y. Chan, J. Chem. Soc., Chem. Commun., 1985, 48.

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